

Clerodane Diterpenes from *Tinospora rumphii*

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Two new diterpenes (**1** and **2**) were obtained from the leaves of *Tinospora rumphii*, along with the known compounds tinotufolin D and vitexilactone. The structures of compounds **1** and **2** were established on the basis of spectroscopic studies.

Tinospora rumphii Boerl. (syn. *T. crispata*; Menispermaceae) is widely distributed throughout the Philippines. Aqueous plant extracts are prescribed in the treatment of stomach trouble, indigestion, diarrhea, and topical ulcers.^{1,2} The leaves in powdered form are used against fever. A preparation with coconut oil is an effective cure for rheumatism and also for flatulence among children. A decoction of the stem is reputed to be an excellent remedy for itches and cancerous wounds. The plant is also used as an antimalarial.^{1,3}

Previous studies on the genus *Tinospora* have led to the isolation of clerodane-type furanoid diterpenes and glucosides.^{4–13} We now report the isolation of two new diterpenes (**1** and **2**) together with the known diterpenes tinotufolin D¹³ and vitexilactone,¹⁴ from the CHCl₃ extract of *T. rumphii*

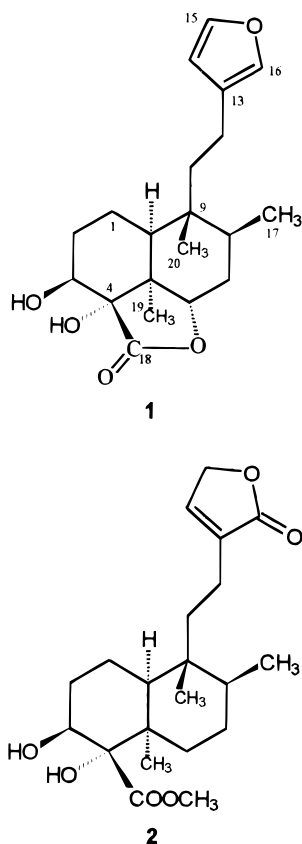


Table 1. ¹³C NMR (δ Values) Data of Compounds **1** and **2** in CDCl₃^a

| position | 1 | 2 |
|------------------|----------|----------|
| 1 | 19.0 | 20.7 |
| 2 | 28.2 | 27.9 |
| 3 | 71.6 | 72.2 |
| 4 | 81.0 | 81.3 |
| 5 | 45.5 | 41.5 |
| 6 | 75.9 | 25.6 |
| 7 | 29.2 | 26.2 |
| 8 | 37.1 | 36.5 |
| 9 | 39.8 | 39.2 |
| 10 | 42.2 | 42.3 |
| 11 | 45.5 | 41.5 |
| 12 | 19.7 | 20.8 |
| 13 | 125.1 | 136.8 |
| 14 | 110.8 | 144.3 |
| 15 | 142.8 | 70.1 |
| 16 | 138.5 | 174.4 |
| 17 | 18.7 | 18.0 |
| 18 | 181.0 | 175.6 |
| 19 | 17.8 | 25.3 |
| 20 | 22.8 | 23.1 |
| OCH ₃ | | 52.8 |

^a Run at 100 MHz; multiplicities determined by a J_{mod} ¹³C experiment.

leaves. The structures of **1** and **2** were elucidated by extensive NMR, IR, UV, and mass spectrometry analysis after separation by Si gel chromatography.

To determine the number of attached protons, a J -modulated spin-echo spectrum for X-nuclei coupled to ¹H was obtained. The J_{mod} ¹³C NMR spectral data of **1** and **2** (Table 1) indicated that both compounds have the same basic diterpene skeleton, with **2** possessing an additional sp³ carbon resonance assigned to a methyl ester. The ¹³C and ¹H NMR (Table 2) spectral data further indicated that **1** contains a β -substituted furan ring, whereas in **2** this functional group is replaced by a butenolide moiety. Three methyl proton signals, two as singlets and one as a doublet, were also common to both compounds.

High-resolution EIMS of **1** gave a molecular ion at m/z 348.1937, with the calculated value for C₂₀H₂₈O₅ being 348.1937. Thus, **1** has an index of hydrogen deficiency of 7. The UV spectrum showed a λ_{max} at 207 nm, supporting the presence of a furan ring.^{8,14–17} The ¹H NMR spectrum of **1** (Table 2) indicated resonances for two carbinyl protons at δ 4.00 (1H, br s) and 5.00 (1H, dd, $J = 9.2, 4.0$ Hz), consistent with the presence of one or more hydroxyl groups (IR ν_{max} 3450, 1129, and 1208 cm⁻¹) and a lactone (IR ν_{max} 1754 and 1258 cm⁻¹), respectively. The COSY NMR spectrum of **1** indicated four isolated spin systems, one of which was a β -substituted furan [δ 6.25 (H-16), 7.35 (H-15), and 7.2 (H-14)]. The remaining three spin systems

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Table 2. ^1H NMR (δ Values) Spectral Data and HMBC Correlations of Compounds **1** and **2** in CDCl_3^a

| position | 1 | HMBC (1) | 2 | HMBC (2) |
|------------------|--------------------|------------------------------|----------------|-----------------------------|
| 1 | 1.52 m, 1.90 m | 2H-2, H-10 | 1.55 m, 1.90 m | |
| 2 | 1.80 m, 1.85 m | 2H-1 | 1.80 m, 1.90 m | |
| 3 | 4.00 br s | | 3.83 br s | |
| 4 | | H-6, H-19 | | H-19 |
| 5 | | H-6, H-10, H-19 | | H-7a, H-19 |
| 6 | 5.00 dd (9.2, 4.0) | 2H-7, H-19 | 1.10 m, 2.50 m | H-19 |
| 7 | 1.85 m, 2.10 m | H-6, H-8, H-17 | 1.35 m, 1.80 m | H-6a, H-17 |
| 8 | 2.05 m | 2H-7, H-10, H-11, H-17, H-20 | 1.60 m | H-17, H-20 |
| 9 | | 2H-7, H-8, H-11, H-17, H-20 | | H-17, H-20 |
| 10 | 1.85 m | 2H-1, 2H-2, H-19, H-20 | 1.75 m | 2H-1, H-8, H-11, H-19, H-20 |
| 11 | 1.50 m, 1.87 m | 2H-12, H-10, H-20 | 1.45 m, 1.95 m | H-12, H-20 |
| 12 | 2.40 (2H) m | 2H-11 | 2.35 (2H) m | H-11, H-14 |
| 13 | | H-12, H-14, H-15 | | H-12, H-14, H-15 |
| 14 | 6.25 dd (1.2, 0.6) | H-12, H-16 | 7.06 br s | H-12, H-15 |
| 15 | 7.35 t (1.2) | H-14, H-16 | 4.76 br s | H-14 |
| 16 | 7.20 br s | H-12, H-14, H-15 | | H-14, H-15 |
| 17 | 1.10 d (5.3) | 2H-7, H-8 | 1.01 d (7.2) | H-8 |
| 18 | | | | H-3, OCH ₃ |
| 19 | 1.25 s | H-6, H-10 | 1.20 s | H-6 |
| 20 | 0.95 s | H-10, 2H-11 | 0.93 s | |
| OCH ₃ | | | 3.85 s | |

^a Run at 400 MHz; *J* values in Hz.

were as follows: the proton at δ 1.52 (H-1a) was coupled to the hydrogen at δ 1.90 (H-1b) and the methylene protons at δ 1.80 (H-2a) and 1.85 (H-2b), which in turn were coupled to the carbonyl proton at δ 4.00 (H-3). The methylene protons at δ 1.85 (H-7a) and 2.10 (H-7b) were coupled to a carbonyl proton at δ 5.00 (H-6) and the methine proton at δ 2.05 (H-8). On the other hand, the methylene protons at δ 1.50 (H-11a) and 1.87 (H-11b) showed cross-peaks with the two-proton multiplet at δ 2.40 (2H-12). The lactone ring was indicated by a carbonyl resonance at δ 181.0. The ^1H and ^{13}C NMR assignments for **1** were determined from their HMQC correlations, and all the connectivities were verified by an inverse long-range heteronuclear experiment optimized for $J = 10$ Hz (Table 2). All long-range correlations observed were consistent with the proposed structure for compound **1**.

The relative stereochemistry of **1** was determined on the basis of NOESY experiments, which indicated key through-space correlations for the proposed framework. The carbonyl proton at δ 4.00 (H-3) was found to be close in space to the hydroxyl protons at δ 2.75 and 3.05. Thus, the carbonyl proton (br s) was assigned to the equatorial position and the hydroxyl protons were assigned to the axial positions. The Me group (δ 1.25, H-19) attached to C-5 was determined as being close in space to the methine proton at δ 1.85 (H-10). Thus, they were found to lie on the same side of **1**, indicating equatorial and axial orientations, respectively. The carbonyl proton at δ 5.00 (H-6) was found to be close to the C-8 Me protons at δ 1.10 (H-17) and the protons at δ 1.90 (H-1b) and 1.85 (H-7a). This indicated that H-6 ($J = 9.2, 4.0$ Hz) and the C-8 Me group are in axial positions. The C-9 Me protons at δ 0.95 (H-20) were close to the C-8 Me protons at δ 1.10 (H-17) and the proton at δ 1.50 (H-1a). This indicates that the C-9 Me group is equatorial. This stereochemistry is supported by comparison with the spectral data for tinotufolin D,¹³ which was also isolated from the *T. rumphii* extract and which could result by dehydration of **1**. Thus, compound **1** was established as (2 α ,3 α ,5 α ,6 β ,7 α ,8 α)-6-[2-(3-furanyl)ethyl]-2 α ,3,4,5,5 α ,6,7,8,8 α ,8b-decahydro-2 α ,3-dihydroxy-6,7,8b-trimethyl-2H-naphtho[1,8-*bc*]furan-2-one.

The structure of **2** was supported by the HREIMS, which gave an $[\text{M} - \text{H}_2\text{O}]^+$ ion at m/z 362.2082; the calculated value for $\text{C}_{21}\text{H}_{30}\text{O}_5$ is m/z 362.2093. Comparison of the ^1H and ^{13}C NMR spectral data of **1** and **2** (Tables 1 and 2)

indicated the presence of a butenolide function¹⁸ [δ 7.06 (1H) and 4.76 (2H)] in **2** instead of a furan ring. The appearance of a methoxyl group at δ 3.85 (3H, s) in **2** and the absence of a carbonyl proton at ca. δ 5.0 in **1** indicated that the lactone in **1** was changed into a methyl ester in **2**. The presence of two carbonyls of the butenolide and ester were apparent from the ^{13}C NMR resonances at δ 174.4 and 175.6 and the IR maxima at 1748, 1712, 1255, and 1229 cm^{-1} . The presence of a butenolide was supported by the UV spectrum, which showed a λ_{max} at 211 nm.¹⁸ The hydroxyl groups in **1** were retained (IR ν_{max} 3450 cm^{-1}). This was supported by the NMR spectral data of **2** (Tables 1 and 2), which retained the carbonyl proton at δ 3.85 and the carbonyl carbons at δ 72.2 and 81.3.

The ^1H and ^{13}C NMR assignments for **2** were verified by HMQC and HMBC experiments (Table 2), and the relative stereochemistry of **2** was established in a manner similar to **1**. The carbonyl proton appeared at δ 3.85 (H-3) as a broad singlet, and thus it was assigned to the equatorial position, with the C-3 and C-4 hydroxyls in axial positions. On the basis of the NOESY experiment, the ester Me was determined as being close in space to the C-5 Me protons, with the latter in turn close to H-10. The C-8 and C-9 Me protons remain close in space. Thus, **2** has the same relative stereochemistry as **1** and tinotufolin D,¹³ and its structure was established as methyl(1 α ,4 α ,5 α ,6 β ,8 α)-5-[2-(3-furan-3-ene-2-one)ethyl]-1,2,3,4,4 α ,5,6,7,8,8 α -decahydro-1,2-dihydroxy-1-naphthalenecarboxylate.

Experimental Section

General Experimental Procedures. Optical rotations were measured on an Optical Activity Ltd. automatic polarimeter. IR spectra were recorded on a Perkin-Elmer 1600 Fourier Transform IR spectrometer and UV spectra on a HP 8452A diode array spectrometer. NMR spectra were recorded on a Bruker Avance 400 in CDCl_3 at 400 MHz for ^1H and 100 MHz for ^{13}C . The high- and low-resolution EIMS were recorded on a Micromass AutoSpec mass spectrometer. Column chromatography was performed with silica gel 60 (70–230 mesh); TLC was performed with plastic-backed plates coated with silica gel F₂₅₄; plates were visualized by spraying with vanillin– H_2SO_4 and warming.

Plant Material. The sample was collected from Antipolo, Rizal, Philippines, in November 1997. It was identified as *Tinospora rumphii* Boerl (Menispermaceae) by Mercedita

Bangis at the Philippine National Museum, and a voucher specimen DLSUCD # 035 is kept at the Chemistry Department of De La Salle University.

Extraction and Isolation. Air-dried leaves of *Tinospora rumphii* (700 g) were ground in an osterizer, then extracted with CHCl_3 (2.8 L) at room temperature for 2 days. The mixture was filtered and the filtrate concentrated in vacuo to afford a crude extract (50 g). This extract was dissolved in EtOH (900 mL), then placed on an ice bath. To the solution was added 4% aqueous $\text{Pb}(\text{OAc})_2$ (900 mL) to precipitate the more polar components.¹⁹ The mixture was then filtered and the filtrate concentrated in vacuo until a mixture of water and an oily residue remained. The concentrate was extracted with CHCl_3 , and the extract was dried with anhydrous Na_2SO_4 , then filtered. The filtrate was concentrated in vacuo to afford the treated extract (10 g), which was fractionated by gravity column chromatography using increasing proportions of acetone in chloroform (10% increments) as eluents. The 10–20% acetone in chloroform fractions were rechromatographed (2 \times) in 15% ethyl acetate in petroleum ether to afford tinotufolin D¹³ (12.5 mg). The 30–50% acetone in chloroform fractions were rechromatographed (3 \times) in dichloromethane–diethyl ether–acetonitrile (9:0.5:0.5) to afford two colorless oils **1** (5 mg) and **2** (15.5 mg) and crystalline vitexilactone¹⁴ (20 mg).

Compound 1: colorless oil; $[\alpha]_{\text{D}}^{30} -32^\circ$ (*c* 0.05, EtOH); IR (KBr) ν_{max} 3450 (br, OH), 1754 (lactone), 1258, 1208, 1129; UV (EtOH) λ_{max} 207 nm (ϵ 3560); ¹H NMR and ¹³C NMR data are listed in Tables 1 and 2; EIMS *m/z* 348 [M]⁺ (14), 254 (66), 218 (34) 173 (21), 191 (66), 149 (100), 121 (58), 110 (73); HREIMS *m/z* 348.1937 [M]⁺ ($\text{C}_{20}\text{H}_{28}\text{O}_5$ requires 348.1937).

Compound 2: colorless oil; $[\alpha]_{\text{D}}^{30} +96^\circ$ (*c* 0.02, EtOH); IR (KBr) ν_{max} 3450 (br, OH), 1748 (lactone), 1712 (ester), 1255, 1229; UV (EtOH) λ_{max} 211 nm (ϵ 3740); ¹H NMR and ¹³C NMR data are listed in Tables 1 and 2; EIMS *m/z* 362 [M]⁺ (15), 325 (48), 309 (71) 267 (33), 251 (86), 247 (100), 235 (57), 123 (54); HREIMS *m/z* 362.2082 [M - H₂O]⁺ ($\text{C}_{21}\text{H}_{30}\text{O}_5$ requires 362.2093).

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References and Notes

- Quisumbing, E. *Medicinal Plants of the Philippines*; Bureau of Printing: Manila, Philippines, 1951; pp 300–301.
- Concha, J. *Philippine National Formulary (Medicinal Plants)*; National Science Development Board: Manila, Philippines, 1982; p 148.
- De Padua, L. S.; Lugod, G. C.; Pancho, J. V. *Handbook on Philippine Medicinal Plants*; Documentation and Information Section, Office of the Director of Research, University of the Philippines at Los Baños: Manila, Philippines, 1981; p 44.
- Fukuda, N.; Yonemitsu, M.; Kimura, T.; Hachiyama, S.; Miyahara, K.; Kawasaki, T. *Chem. Pharm. Bull.* **1985**, *33*, 4438–4444.
- Fukuda, N.; Yonemitsu, M.; Kimura, T. *Chem. Pharm. Bull.* **1986**, *34*, 2868–2872.
- Fukuda, N.; Yonemitsu, M.; Kimura, T. *Liebigs Ann. Chem.* **1993**, *33*, 491–495.
- Martin, T. S.; Ohtani, K.; Kasai, R.; Yamasaki, K. *Phytochemistry* **1995**, *40*, 1729–1736.
- Hanuman, J. B.; Bhatt, R. K.; Sabata, B. *J. Nat. Prod.* **1988**, *51*, 197–201.
- Gangan, V. D.; Pradhan, P.; Sipahirmalani, A. T.; Banerjee, A. *Phytochemistry* **1995**, *39*, 1139–1142.
- Maurya, R.; Wazir, V.; Tyagi, A.; Kapil, R. S. *Phytochemistry* **1995**, *38*, 659–661.
- Gangan, V. D.; Pradhan, P.; Sipahirmalani, A. T.; Banerjee, A. *Phytochemistry* **1994**, *37*, 781–786.
- Bhatt, R. K.; Sabata, B. K. *Phytochemistry* **1989**, *28*, 2419–2422.
- Fukuda, N.; Yonemitsu, M.; Kimura, T.; Isobe, R.; Komori, T. *Liebigs Ann. Chem.* **1994**, 755–759.
- Kondo, Y.; Sugiyama, K.; Nozoe, S. *Chem. Pharm. Bull.* **1986**, *34*, 4829–4832.
- Hanuman, J. B.; Bhatt, R. K.; Ahmad, S.; Sabata, B. K. *Phytochemistry* **1986**, *25*, 1677–1680.
- Atta-Ur-Rahman; Safdar Ali, S.; Ahmad, S.; Chaudhary, M. D. *Phytochemistry* **1992**, *31*, 3155–3157.
- Bhatt, R. K.; Hanuman, J. B.; Sabata, B. K. *Phytochemistry* **1988**, *27*, 1212–1216.
- Chan, W. R.; Taylor, D. R.; Bodden, R. L. *Tetrahedron* **1971**, *27*, 5081–5091.
- Padolina, W. G.; Yoshioka, H.; Nakatani, N.; Mabry, T. J.; Monti, S. A.; Davis, R. E.; Cox, P. J.; Sim, G. A.; Watson, W. H.; Wu, I. B. *Tetrahedron* **1974**, *30*, 1161–1170.

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